Sample Questions for Chem 161 Placement Exam

1.
$$5x - 3(x - 6) + 2(y - 4) =$$

(1) 2x + 2y - 26 (2) -3x + 2y + 20

(3) 2x + 2y + 10

 $(4) -3x + 2y + 2 \qquad (5) 2x + 2y - 10$

2. If $5^x = 8$ then x =

(1) $\frac{8}{5}$ (2) $\frac{5}{8}$ (3) $\log_5 8$ (4) $\log_8 5$ (5) $\log_{10} \left(\frac{8}{5}\right)$

The number 3.90 x 10⁹ has how many significant figures? 3.

(1) 2

(2) 3

(3) 4

(4) 9

(5) 13

4. Which bond is most polar?

(1) H-H

(2) H-C

(3) H-N

(4) H-F

(5) H-Cl

For the isotope ¹⁷O², the number of protons, neutrons, and electrons is 5.

(1) 6 protons, 9 neutrons, and 6 electrons

(2) 8 protons, 8 neutrons, and 8 electrons

(3) 8 protons, 17 neutrons, and 8 electrons

(4) 8 protons, 9 neutrons, and 10 electrons

(5) 7 protons, 10 neutrons, and 10 electrons

6. Barium hydroxide dissolves as follows.

$$Ba(OH)_2(s) \rightarrow Ba^{2+}(aq) + 2OH^{-}(aq)$$

How many moles of Ba(OH)₂ are required to make 1.0 L of solution in which the OH concentration is 0.050 M?

(1) 0.025

(2) 0.050

(3) 0.10

(4) 2.0

(5) 0.15